

ESCAPE

An Expert System for Consequence Analysis using a Personal Computer

References:

- Riikonen, K. et al., 1992. Finnish Meteorological Institute, Publ. Air Quality 13.
- http://www.fmi.fi/research_air/air_55.html

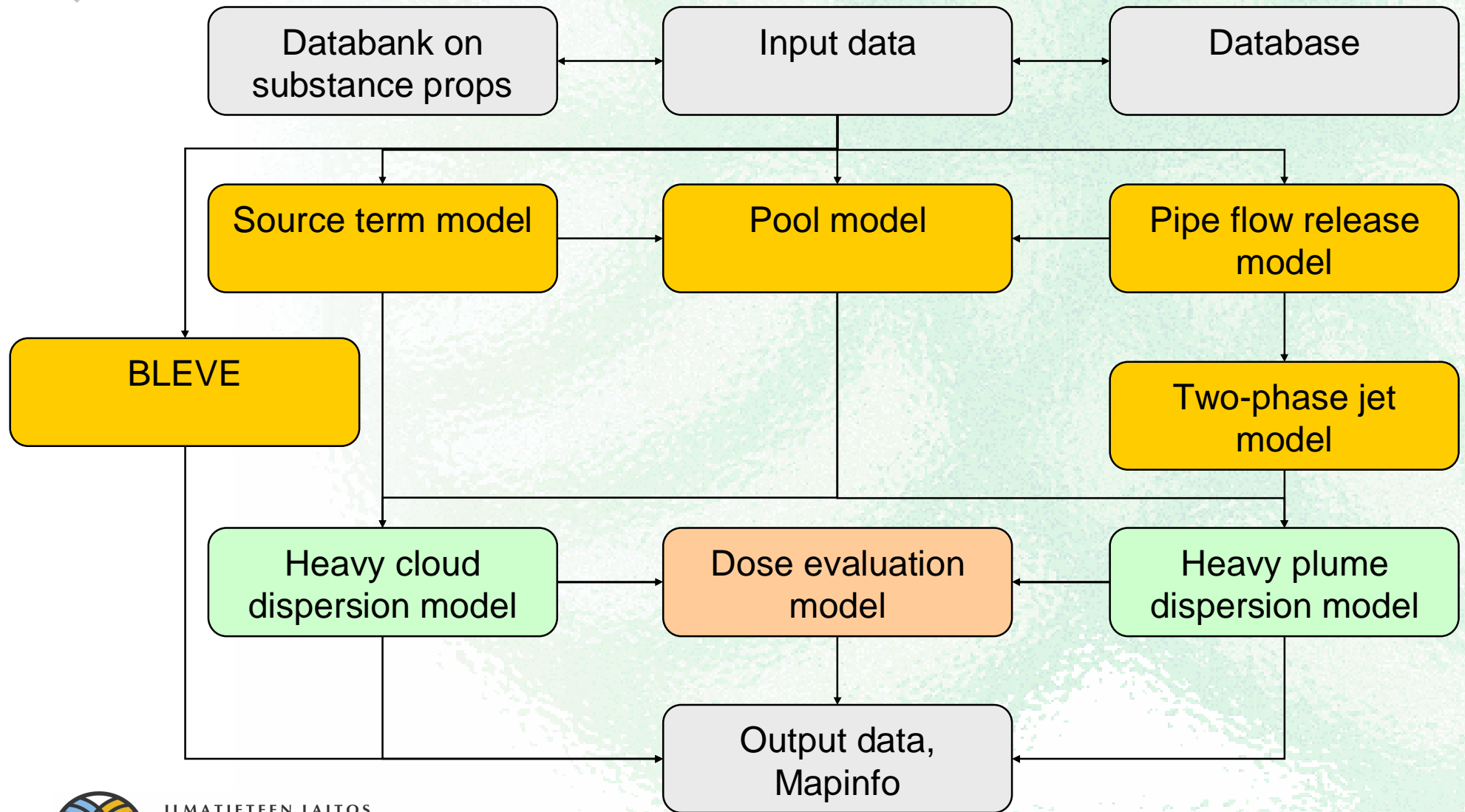


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ESCAPE



The model applies for the following substances

Acetone	Chlorine	Methane
Acetonitrile	Cumene	N-butane
Acrylic acid	Ethane	N-pentane
Acrylonitrile	Ethylene oxide	Phosgene
Ammonia	Freon 12	Propane
Benzene	Hexane	Propylene oxide
Bromine	Hydrogen cyanide	Sulphur dioxide
Butadiene	Hydrogen fluoride	Sulphur trioxide
Butane	Hydrogen sulphide	Toluene
Carbon disulphide	Iso-Butane	Vinyl chloride
Carbon tetrachloride	Methanol	



Effects of ammonia

Concentration mg/m ³	Time of influence	Symptoms	Injuries
3.5 - 35		Noticeable odour	
70		Light irritation of nose	Irritation of mucous membrane
200 - 350	0.5 - 1 h	Irritation of throat, eyes and nose, cough	Light injuries
500		Flowing of tears	Rather serious injuries
1200	0.5 h	Damages in lugn	Serious injuries
2500	0.5 - 1 h	Swelling of the throat	May be lethal
3500 - 7000	10 - 15 min	Swelling of the throat	Death



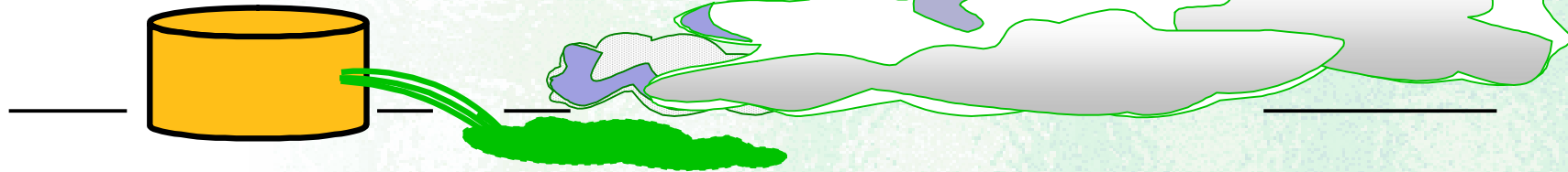
Effects of chlorine

Concentration mg/m ³	Time of influence	Symptoms	Injuries
1.5 - 9		Light odour	
9	8 - 10 h	Noticeable odour	Irritation of the mucous membrane
30	2 h	Coughing	Light injuries
150	20 min	Suffocating cough	Rather difficult injuries
300	10 min	Suffocation cough	Rather difficult injuries
900	3 min		Serious injuries
3000		Suffocation	Death

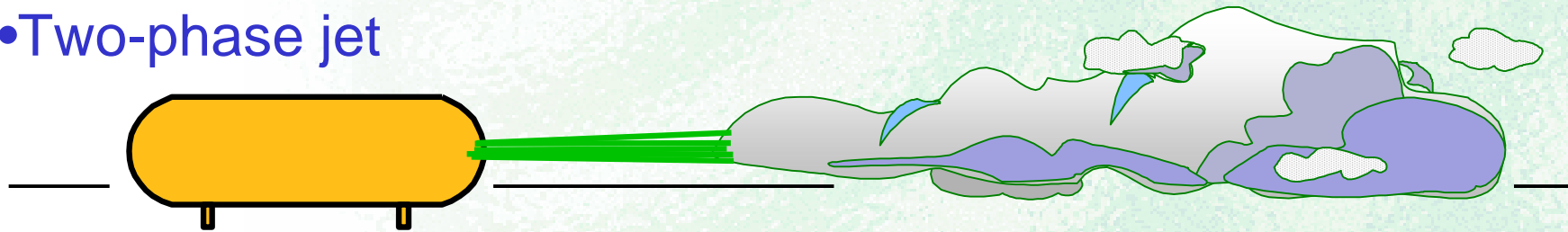


Various source terms

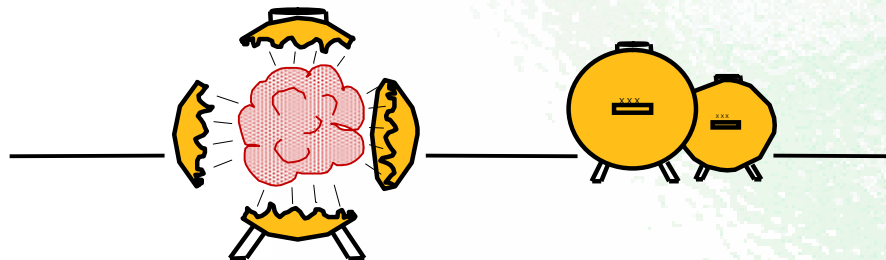
- Pool formation



- Two-phase jet



- Catastrophic failure



Ref.: Webber, 1992



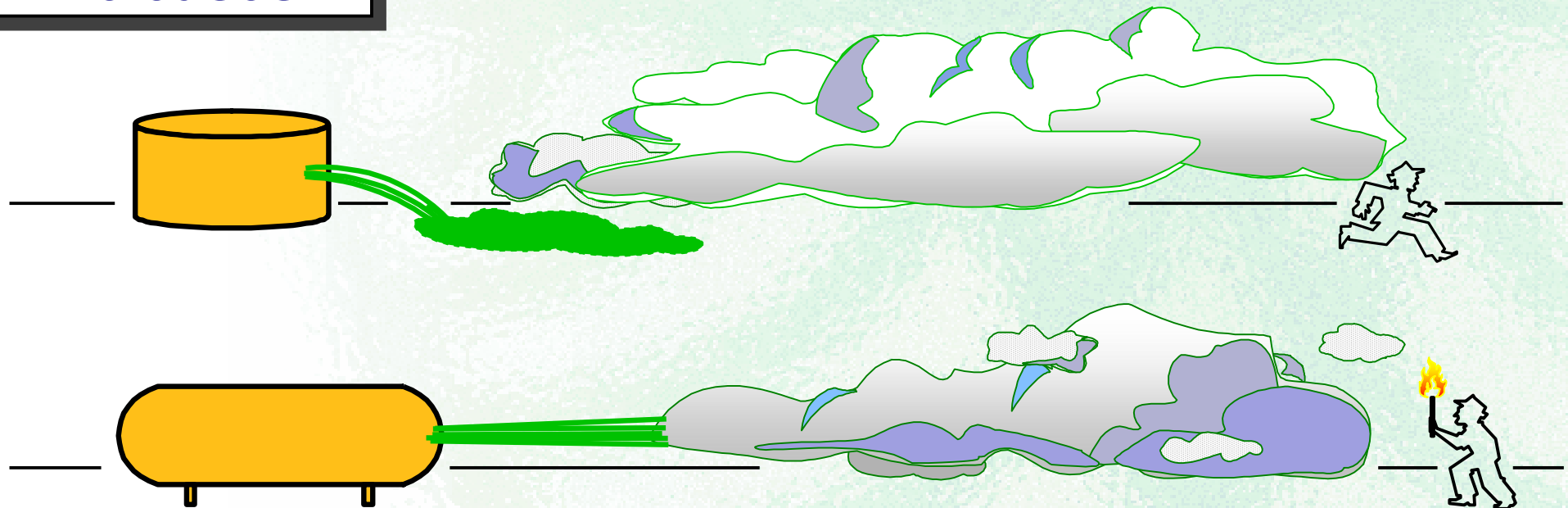
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Source term

Effects

Dispersion

Releases

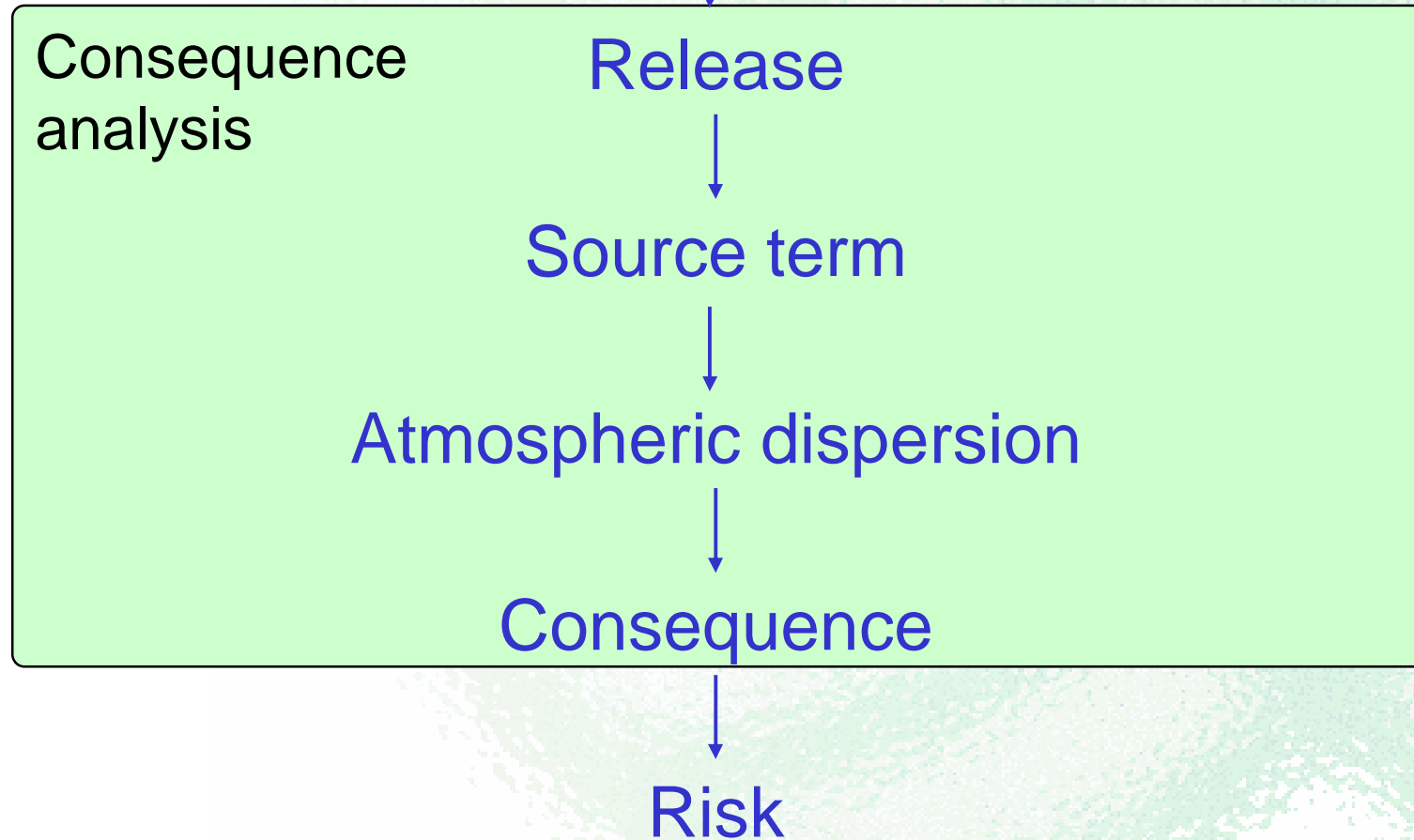


The model ESCAPE evaluates releases, atmospheric dispersion and consequences of chemical accidents (Fig.: Webber, 1992).



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The propability of accident



Application of models and numerical results

- emergency response planning
- hazard and risk analysis
- inspection and surveillance of industrial installations
- analysis of past accidents

The model results are most useful in preparing in advance for accidents.



The state-of-the-art of source term models

1. Liquid pool models
 - fairly good models available
2. Two-phase jet models
 - reasonably good models available
3. Source terms of catastrophic releases
 - the few suggested models not realistic
4. "Semi-instantaneous" releases
 - no models available



The state-of-the-art of heavy gas dispersion models

The atmospheric dispersion of a one-phase gas cloud over flat terrain in terms of time-averaged concentrations is generally well understood.

Predictive techniques are sufficiently accurate for hazard assessment purposes.

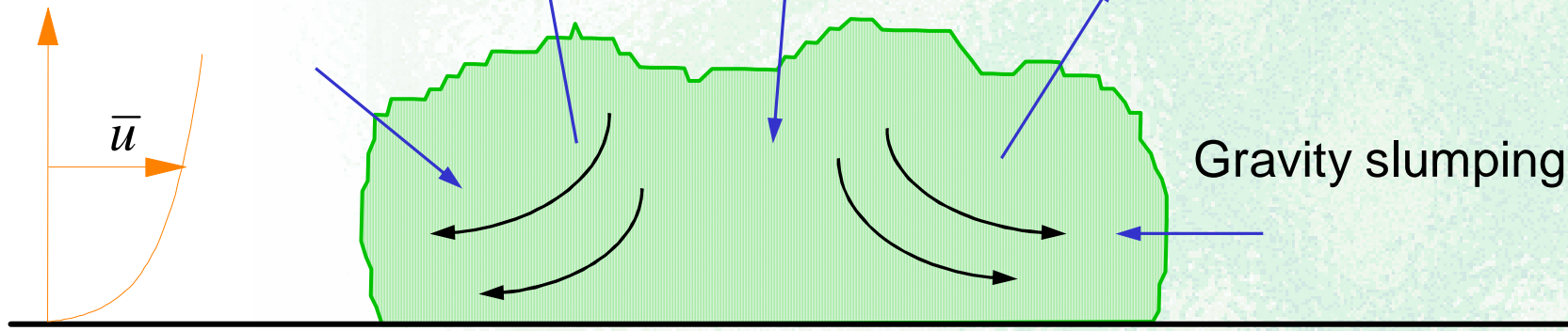


HEAVY GAS DISPERSION MODEL

Entrainment of air:

- atmospheric turbulence
- internal turbulence of the cloud
- mechanical entrainment

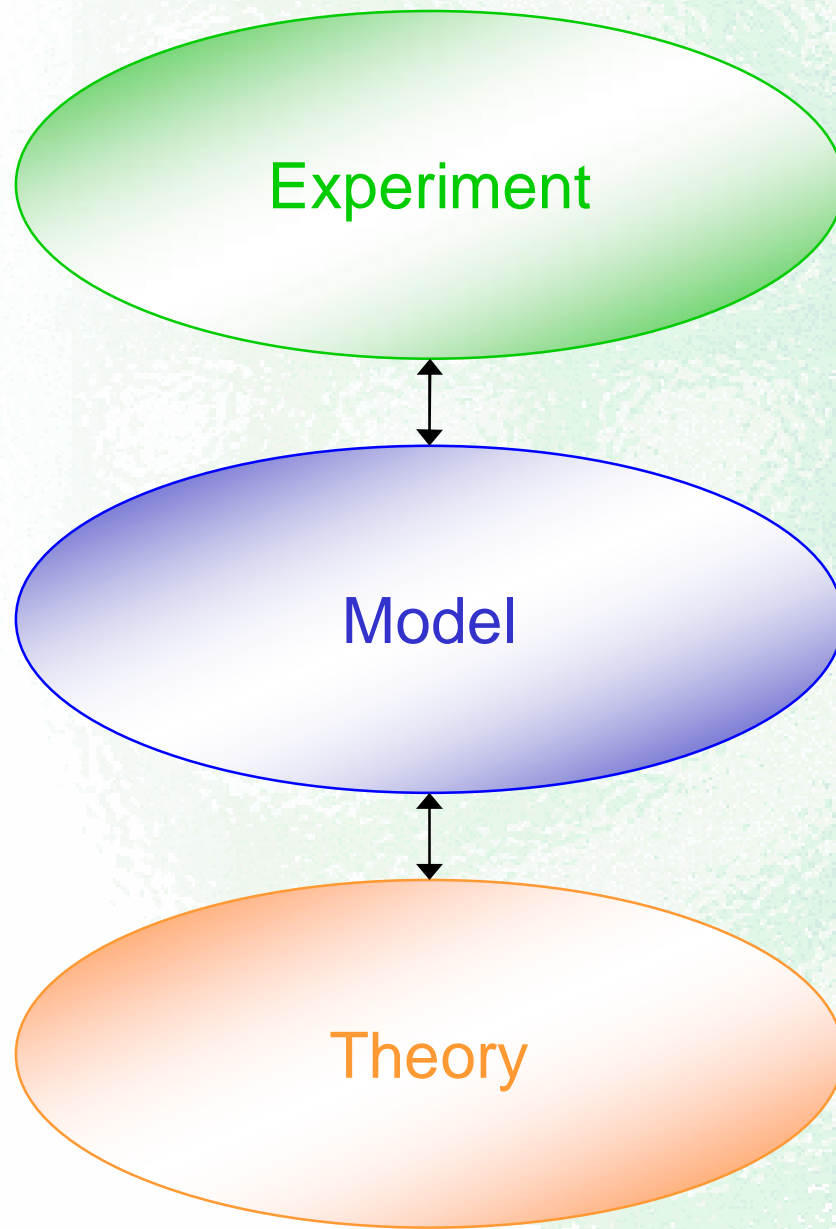
Transport of the cloud



Thermodynamical effects:

- phase transitions
- heat flux from the ground





Refinement of the model

- including a databank of the physico-chemical properties of substances
- an easy-to-use introduction of new substances into the model
- refinement of heavy gas dispersion models
- improvement of the flexible and versatile use of the model
- improvement of graphical output
- further validation against experimental data



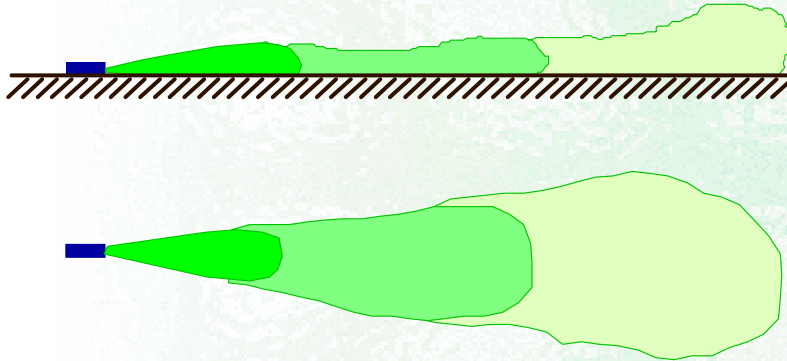
Future research on heavy gas dispersion modelling

- thermodynamical effects
- the influence of complex terrain and obstacles
- statistical fluctuations
- time-dependent releases

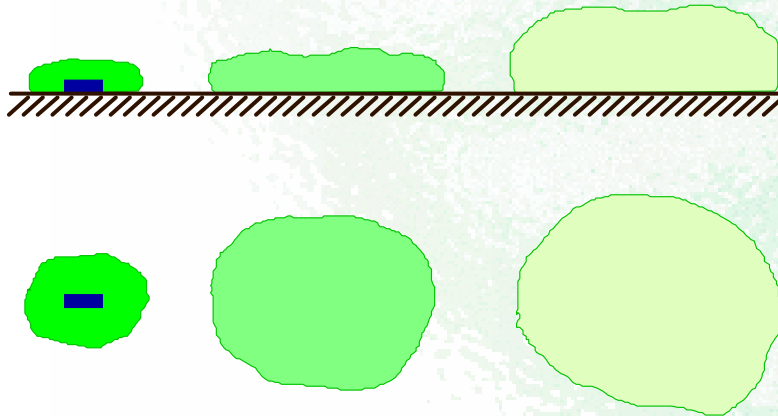
Validated tools of assessment are also needed for non-expert users.



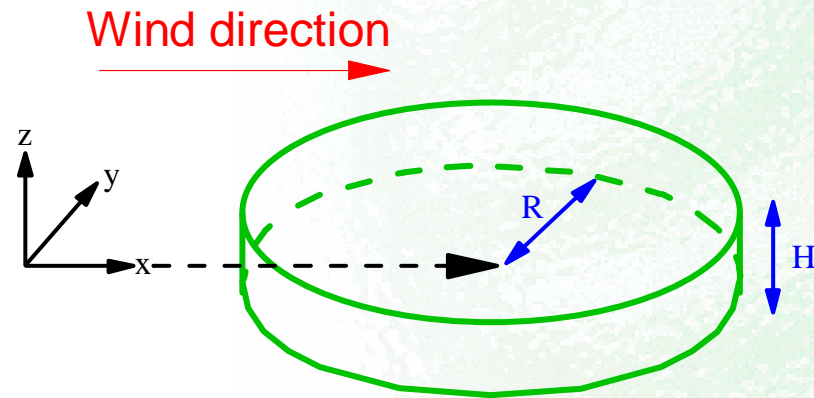
Plume



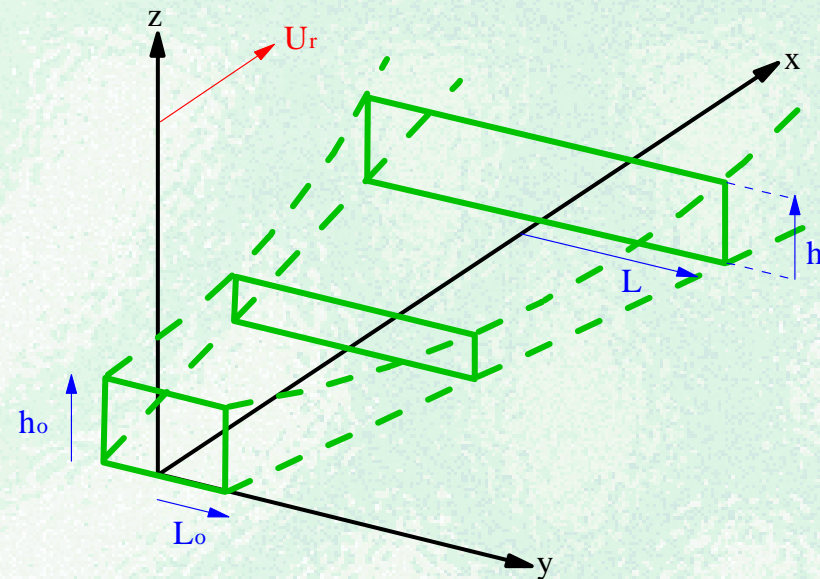
Cloud



Box model cloud representation



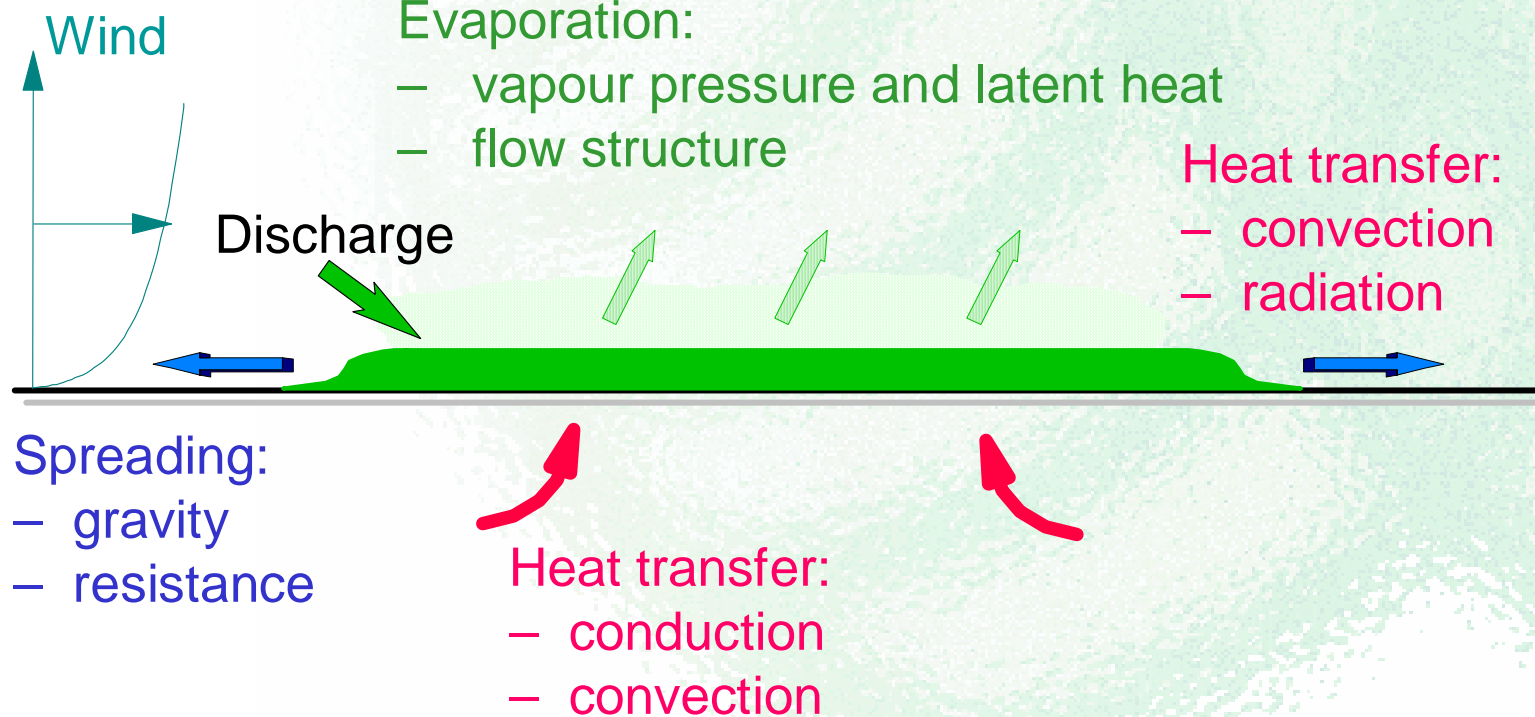
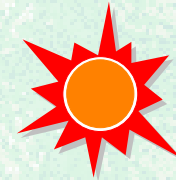
Cloud shape for instantaneous release



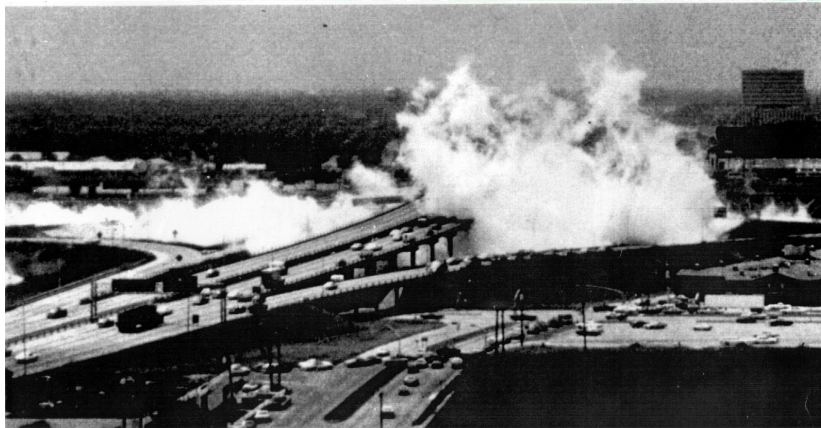
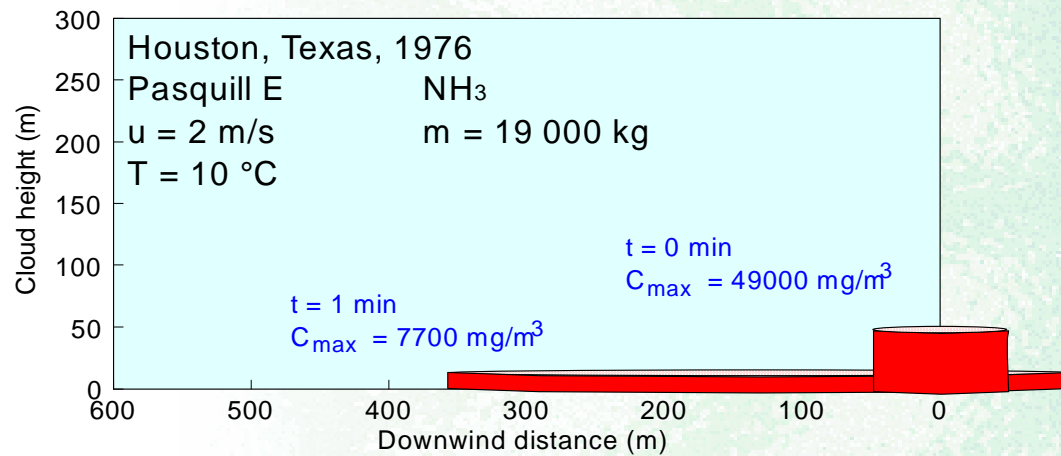
Plume shape for continuous release



A model for liquid pool evaporation and spreading



Comparison of the predictions of ESCAPE and the visual observations



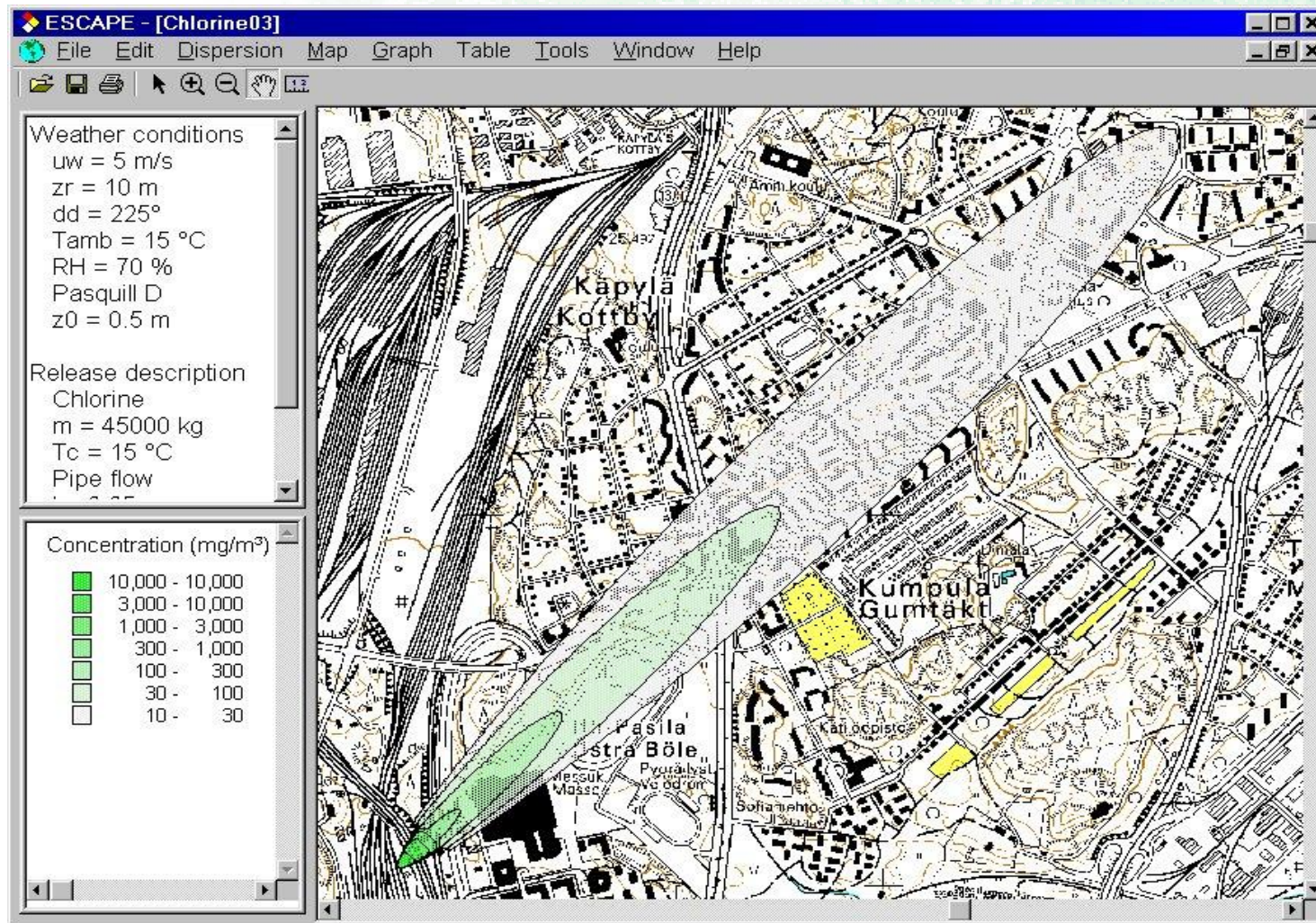
A photograph taken about one minute after the Houston tanker crash (Fryer and Kaiser, 1979).



The results of ESCAPE

- Dispersion diagram
- The height of gas cloud or plume
- The average concentration
- The temperature of gas cloud or plume
- The density of gas cloud or plume
- Tabulated results of concentration
- Total dose
- Dose in a minute
- Tabulated results of doses
- Toxicology and flammable levels
- BLEVE (blast wave dispersion diagram)

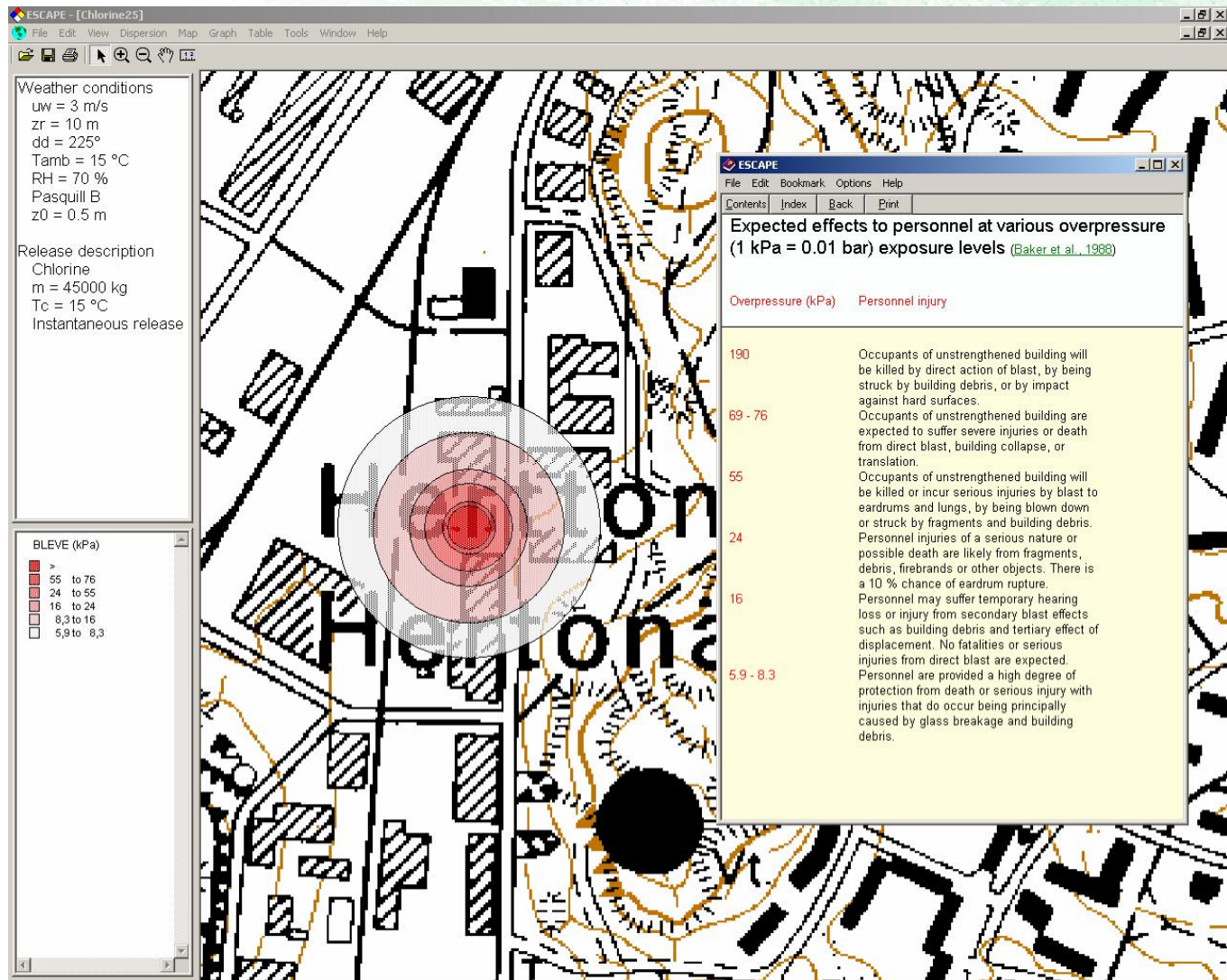




Example output of ESCAPE. (map (©) Genimap Oy).



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Example results of the Win - ESCAPE application: BLEVE formed in a conceived container breakage of liquified chlorine (map (©) Genimap Oy).



ESCAPE Users

City of Helsinki Environment Centre, Finland

Ekokem, Finland

Environment Protection Agency, Lithuania

*Finnish Meteorological Institute (FMI)

*Inspecta, Finland

Institute of Physics of Vilnius, Lithuania

*Ministry of the Interior, Finland

∅ Rescue services (22)

Neste, Finland

Radiation and Nuclear Safety Authority (STUK), Finland

State Pollution Control Authority (SFT), Norway

The Finnish Defence Forces

Toray Plastics Malaysia Sdn. Bhd.

University of Applied Sciences (EVTEK and HAMK), Finland

University of Hamburg, Germany



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*financier

More information

- http://www.fmi.fi/research_air/air_55.html
- Download: <http://escape.fmi.fi/>
(license required!)

Thank you for your attention !



Model equations (POOL)

Volume discharged: $\frac{dV_d}{dt} = S$

Pool volume: $\frac{dV}{dt} = S - WA$

Pool temperature: $c\rho V \frac{dT}{dt} = A\{Q - \rho WL\} + \{T_s - T\}c\rho S$

Mass vaporised: $\frac{dM}{dt} = \rho WA$

$$A = \pi r^2$$

$$\rho W = X(T)u_* \hat{W}(u_* z_o / \nu, Sc, R/z_o, p_s(T)/p_a)$$

$$Q(T, t) = \text{conducted} + \text{convected} + \text{radiated}$$

W = evaporation rate

A = area of pool

c = spec. heat capacity of contaminant

r = liquid density of contaminant

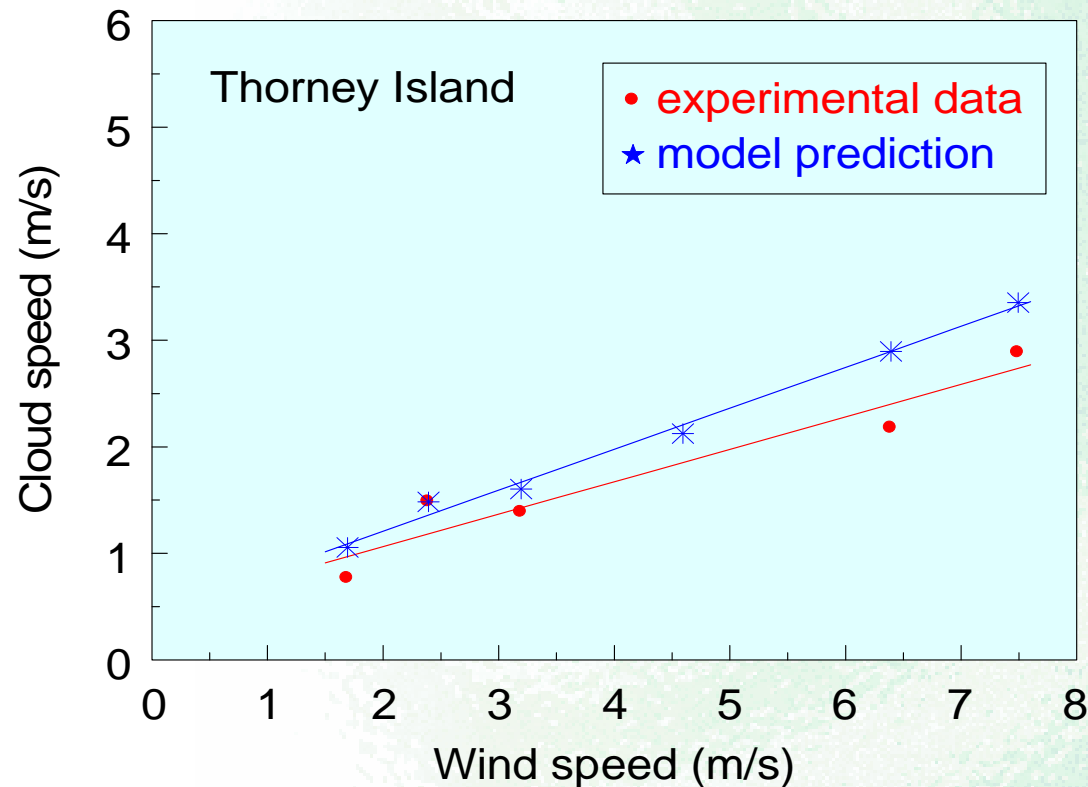
Q = heat flux density into the pool

L = latent heat of evaporation

T_s = temperature of source discharge



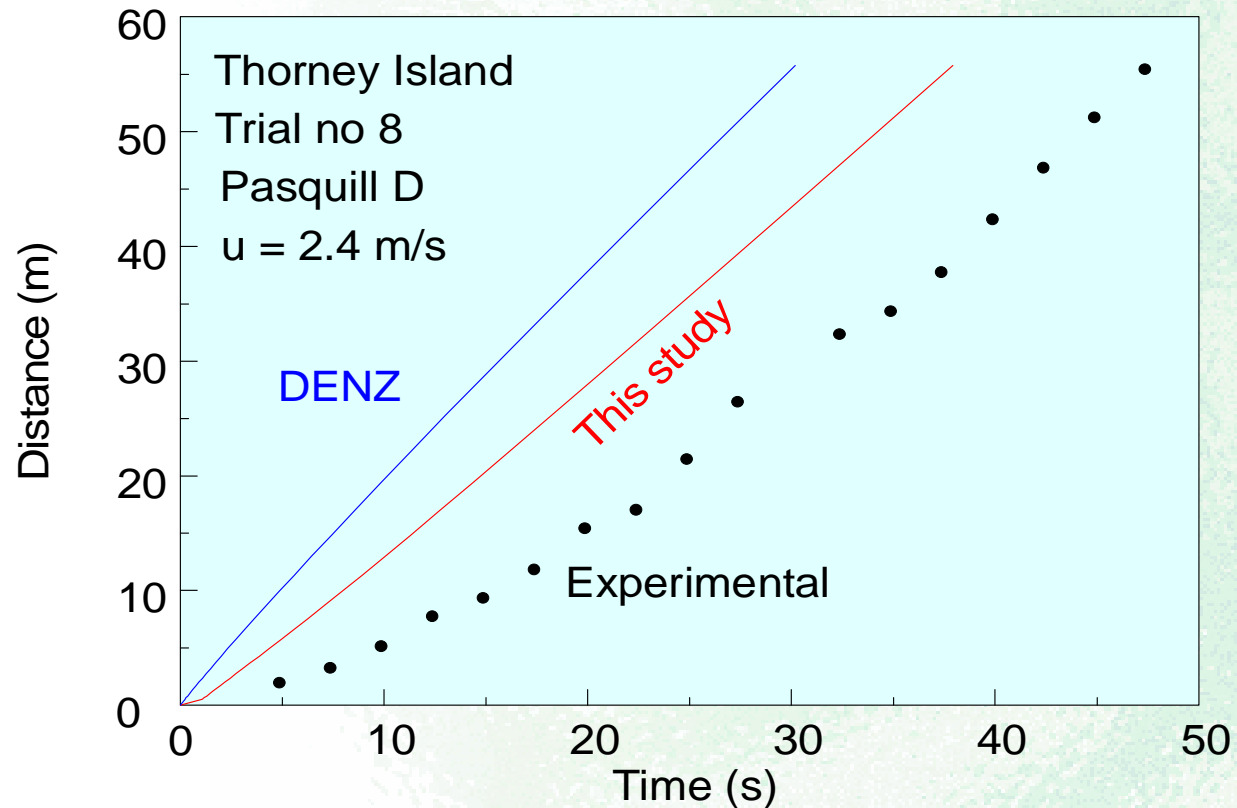
Validation of DENZ model (4/4)



The computed and measured cloud speeds against the measured wind speed at 10 m height (Kukkonen and Nikmo, 1992. J.Hazard.Mater. 31, 155-176)



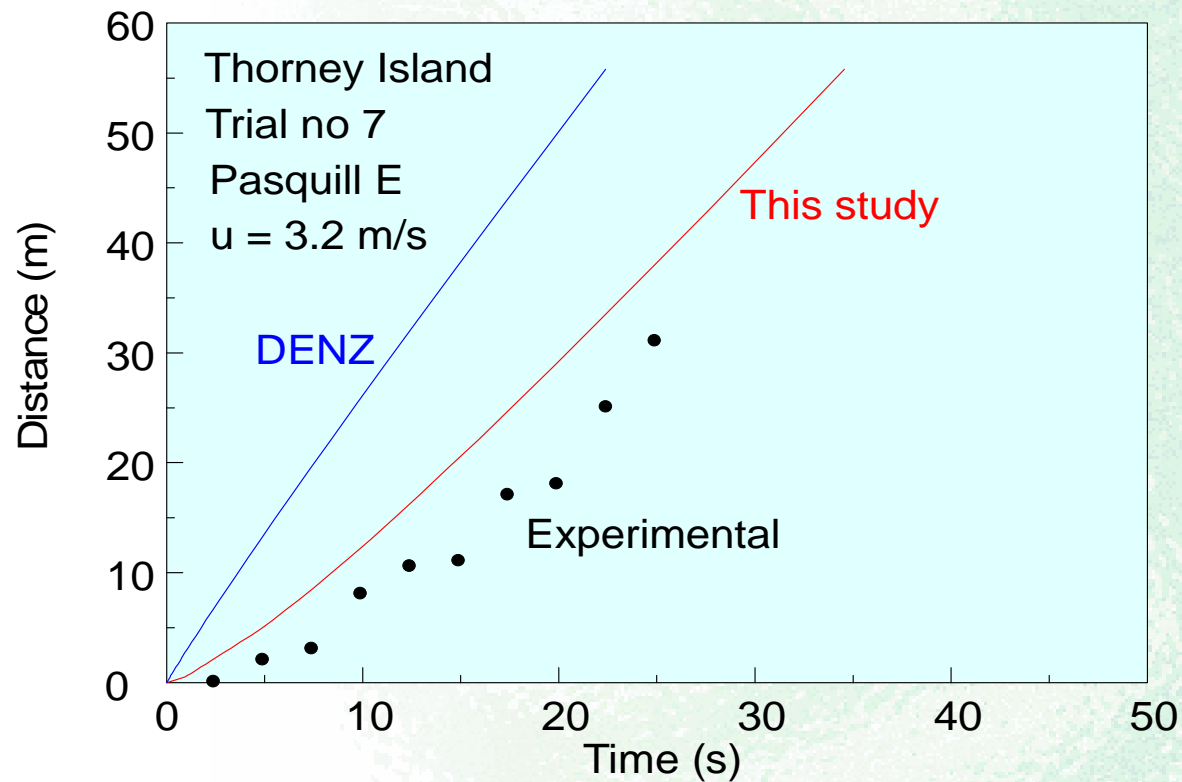
Validation of DENZ model (3/4)



The distance of the cloud centroid from source against time (Kukkonen and Nikmo, 1992. J.Hazard.Mater. 31, 155-176).

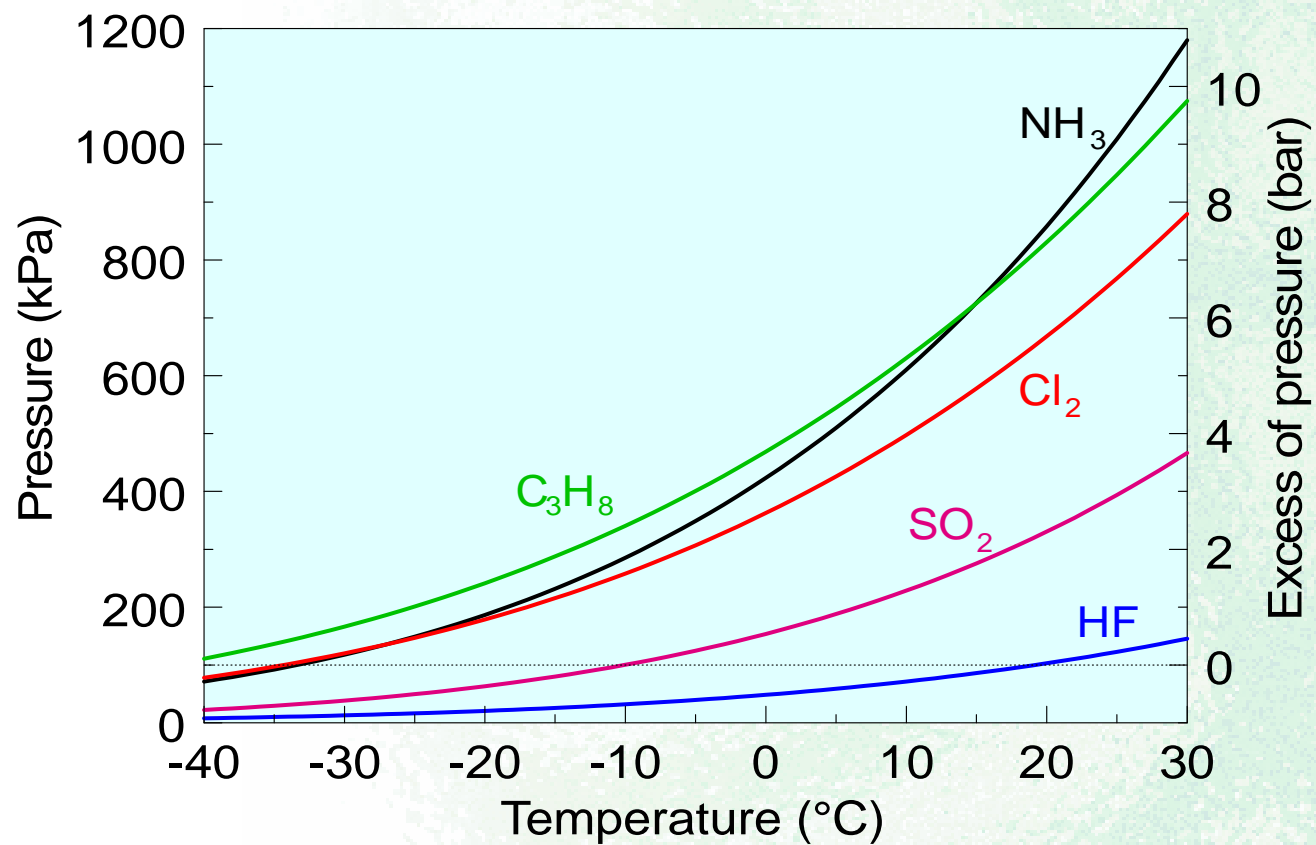


Validation of DENZ model (2/4)



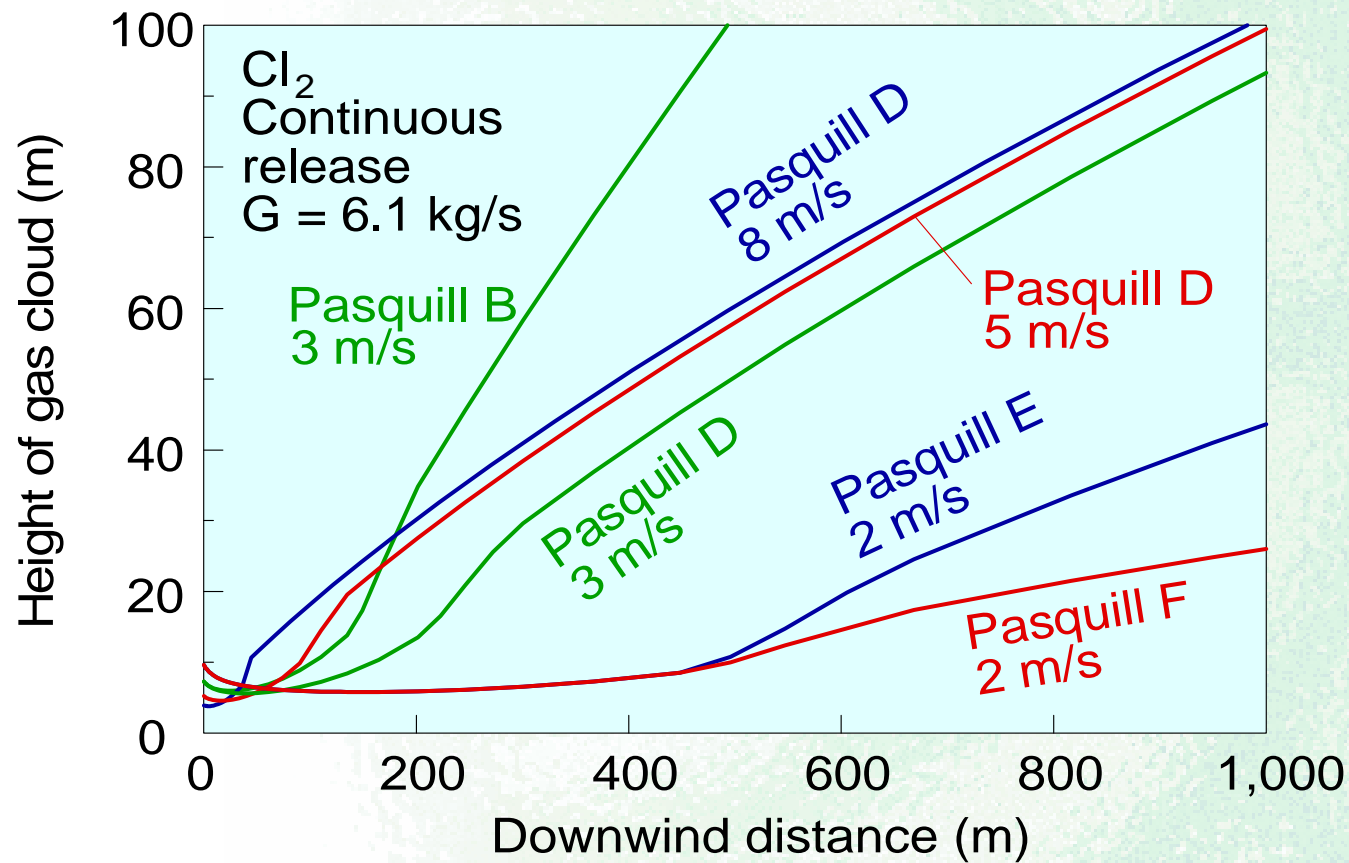
The distance of the cloud centroid from source against time (Kukkonen and Nikmo, 1992. J.Hazard.Mater. 31, 155-176).





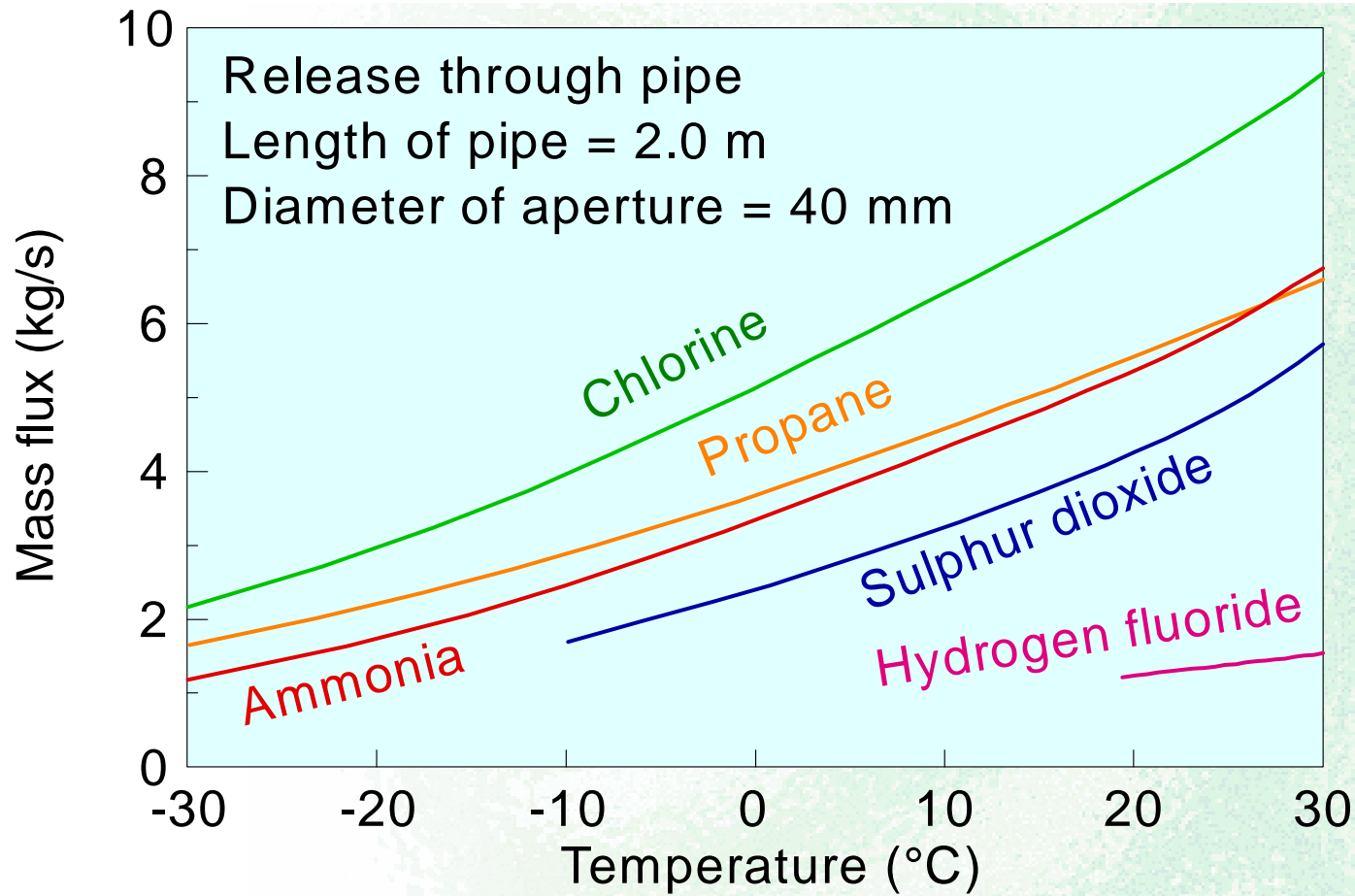
The saturation vapour pressure for some toxic and flammable substances.





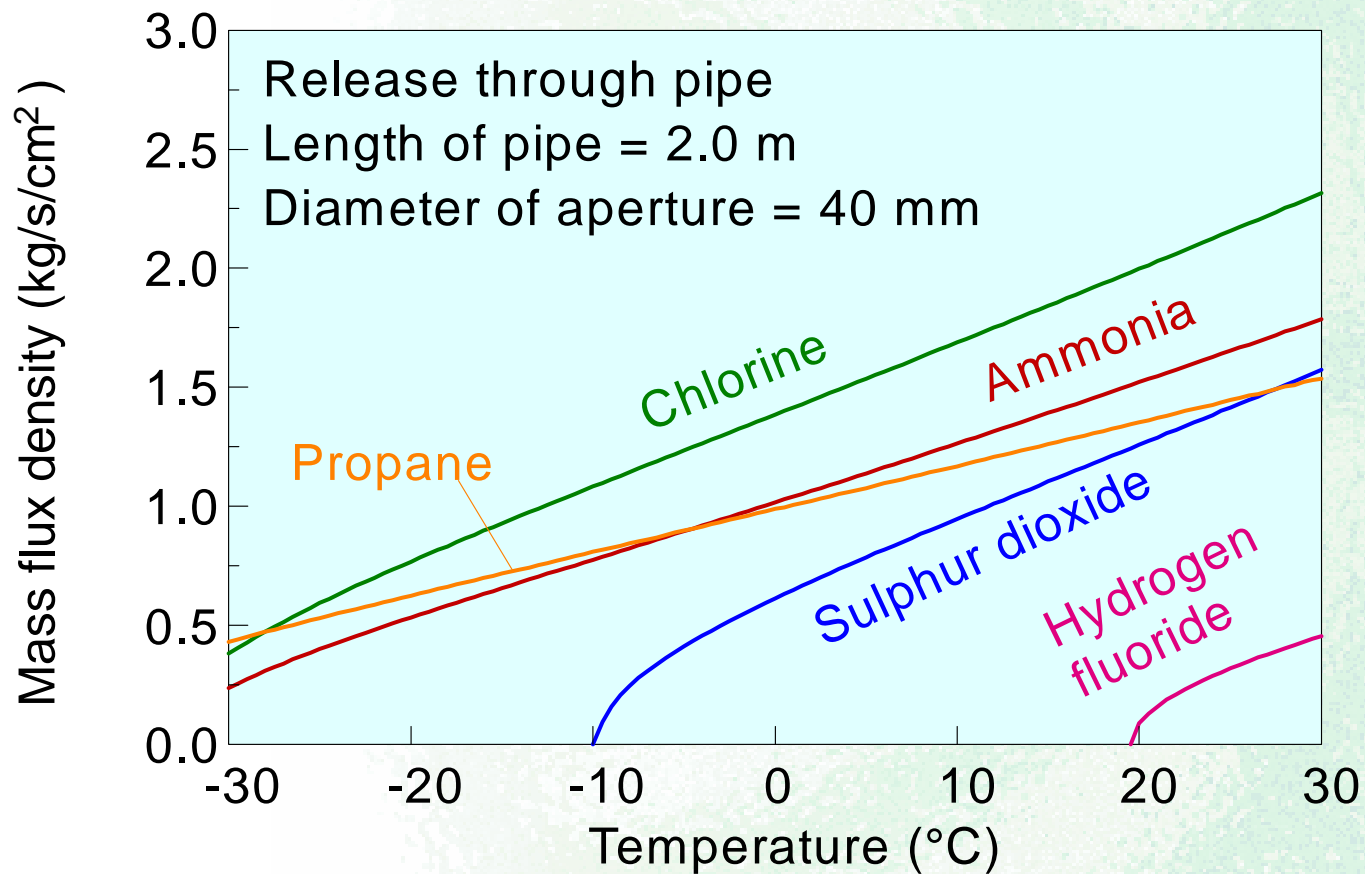
The predicted height of the gas cloud in varying meteorological conditions (Kukkonen and Savolainen, 1988. Publ. on Air Quality 4, Finnish Meteorological Institute).





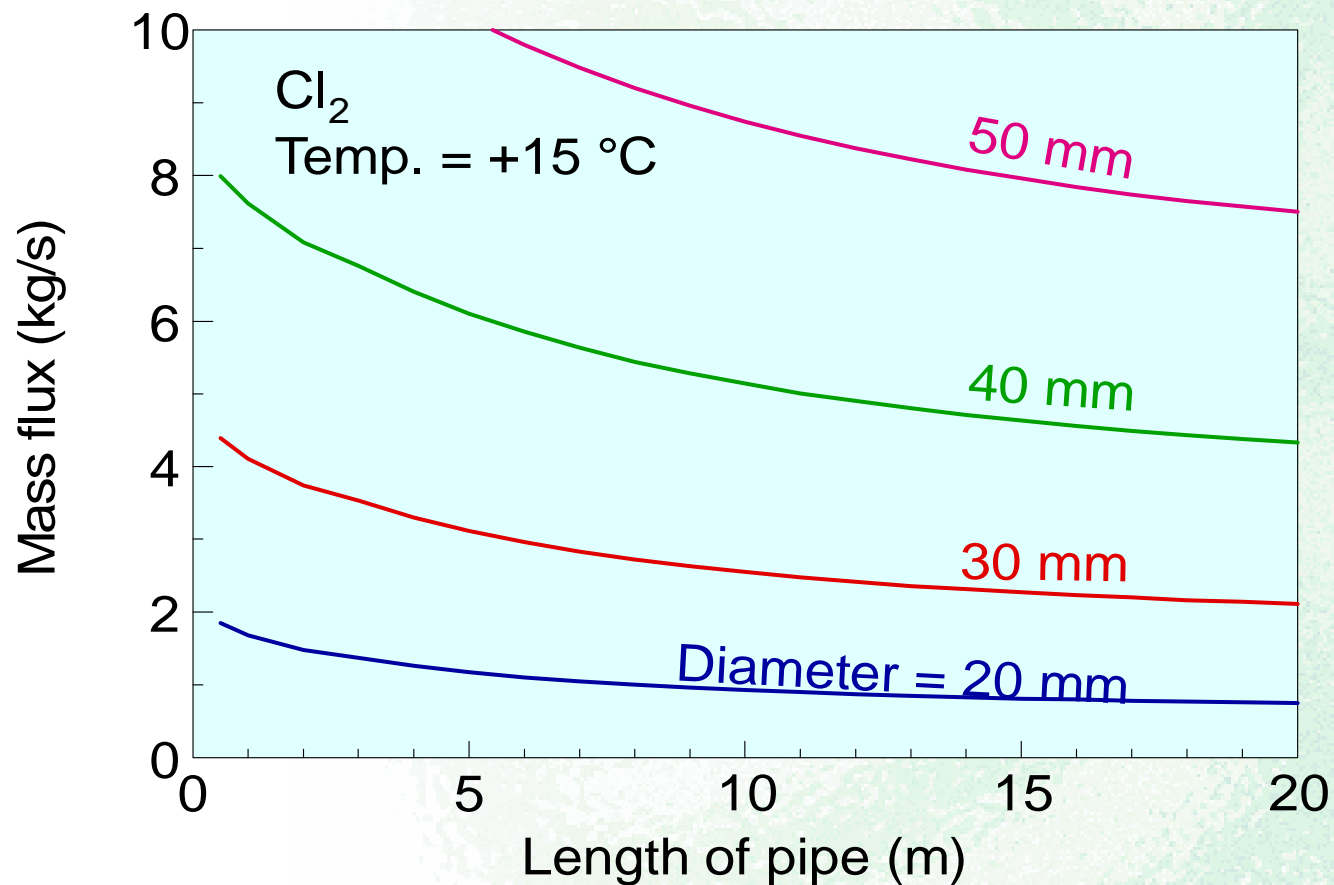
The mass flux through a pipe from the liquid space of the container against the storage temperature (Kukkonen, 1990. Dissertations No 34, The Finnish Society of Sciences and Letters).





The mass flux through a pipe from the liquid space of the container against the storage temperature (Kukkonen, 1990. Dissertations No 34, The Finnish Society of Sciences and Letters).

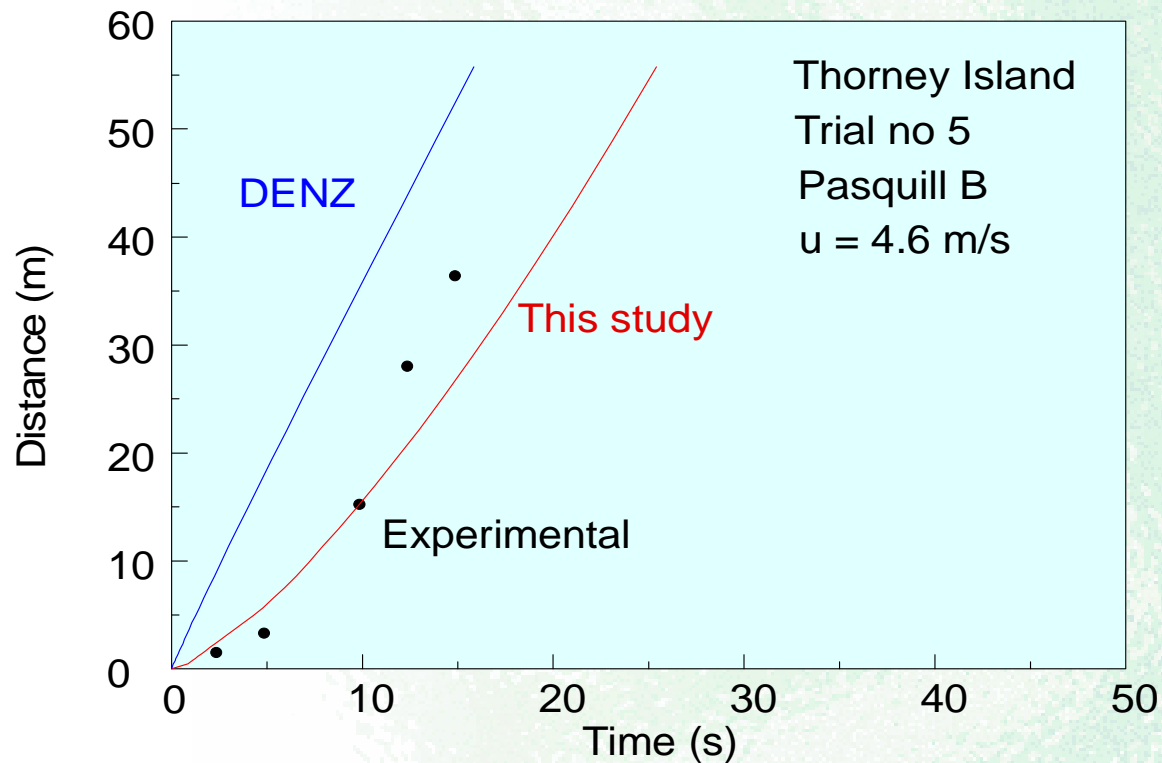




The mass flux through pipe from the liquid space of the container against the length of pipe (Kukkonen, 1990. Dissertations No 34, The Finnish Society of Sciences and Letters).



Validation of DENZ model (1/4)



The distance of the cloud centroid from source against time (Kukkonen and Nikmo, 1992. J.Hazard.Mater. 31, 155-176).

